

1 Methane in sulphides from gold-bearing deposits, Britain and Ireland

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7 Abstract

8 The direct measurement of gases trapped in sulphide minerals shows that samples from gold-
9 bearing deposits in Britain and Ireland are anomalously rich in methane. Chalcopyrite samples in
10 deposits of Palaeozoic age sited in Neoproterozoic basement (Caledonides greenschist), Munster
11 Basin (Variscides greenschist) and the Carboniferous cover (diagenetic) were compared using mass
12 spectrometry of cold-crushed gases. All host sequences contain sources of organic matter. The
13 content of non-aqueous gas is greater in both sets of greenschist-hosted deposits than in the
14 diagenetic-hosted deposits. However, chalcopyrite accompanying gold in the Neoproterozoic is
15 methane-rich, but in the low-gold Munster Basin it is methane-poor. These gas data from opaque
16 minerals complement fluid inclusion data from gangue minerals, and add support to models for the
17 involvement of organic species in orogenic gold mineralization.

18 Introduction

19 Models for mineralization in (meta-)sedimentary rocks commonly include gases within the
20 mineralizing fluids, especially carbon dioxide, hydrogen sulphide and methane, which can influence
21 solute transport and deposition (e.g. Landis & Hofstra 1991; Wilkinson 2001). Some supporting
22 evidence is provided by fluid inclusion compositional data, but this is often limited to the dominant
23 gas (usually CO₂) and to translucent minerals. An alternative approach is the analysis of gases
24 released from fluid inclusions by crushing samples into a mass spectrometer at ambient temperature
25 (Blamey 2012), which can measure multiple gases and from opaque minerals. The gases entrained in
26 minerals include species liberated from the thermal alteration of organic matter, which could
27 include variable mixtures of methane, higher hydrocarbons, hydrogen and carbon dioxide (Blamey
28 2012, Suzuki et al 2017).

29 One model for orogenic gold deposits involves the role of methane for complexing and precipitating
30 gold (Naden & Shepherd 1989; Gaboury 2013; Goldfarb & Groves 2015; Jébrak et al. 2018). This was
31 tested using deposits from Britain and Ireland (Fig. 1) by assessing if the compositions of gases in
32 gold deposits are distinct among samples in three distinct settings. As far as possible all samples
33 are chalcopyrite of Palaeozoic age, to minimize variables, from the Neoproterozoic of Scotland and
34 Northern Ireland, the Munster Basin of southwest Ireland, and widely distributed deposits attributed
35 to Mississippi Valley Type (MVT). Chalcopyrite was used because it occurs in a wide range of host
36 lithologies and redox environments, but lacks the prevalent cleavage found in galena and sphalerite
37 that is conducive to the loss of entrained gases. We avoid pyrite because it is commonly zoned, and
38 thus a single crystal can represent a range of entrapment conditions. The abundance of samples
39 from the three categories reflects their relative distribution: The most widely distributed style of

40 mineralization in Britain and Ireland is limestone-hosted vein deposits in the Carboniferous,
41 considered to be Mississippi Valley Type (MVT) (Ixer 1986; Ixer & Vaughan 1993). The Carboniferous
42 rocks are in the oil window. In addition to the deposits within the Carboniferous, sulphide veins in
43 the sub-Carboniferous basement at Ballygrant, Leadhills, Conlig and Caldbeck are attributed to the
44 lower part of the same mineralizing system (Anderson et al. 1989; Lowry et al. 1991; Moles & Nawaz
45 1996; Baron & Parnell 2005). These deposits occur particularly in dolostone as fillings of open space,
46 in collapse breccias and as replacements of the carbonate host rock. They are epigenetic, emplaced
47 after the lithification of the host rock. Fluid inclusion data indicate an origin from saline basinal
48 brines at temperatures in the range 75 to 200 °C. Sulphur isotope data typically indicate thermogenic
49 sulphate reduction (Thom & Anderson 2008). The Carboniferous-hosted set includes a sample from
50 Navan, which is of exhalative rather than MVT origin. In the southwest, the Devonian-Carboniferous
51 of the Munster Basin, adjacent to the Variscan orogenic front, experienced limited metamorphism to
52 greenschist facies and contains copper ores (Wen et al. 1996; Spinks et al. 2016). Sediment-hosted
53 copper mineralization occurs within reduced grey-green-purple sandstone, representing diagenetic
54 reduced zones. Disseminated chalcopyrite, chalcocite and bornite occur interstitially between pores.
55 Sulphur isotope compositions indicate mineralization by microbial sulphate reduction (Wen et al.
56 1996). Copper mineralization also occurs as veinlets within mineralized red beds, and in E-W
57 trending quartz-barite veins up to 2 m wide, stratigraphically above the diagenetic sediment-hosted
58 copper deposits. The minor and major vein mineralization is thought to be the result of
59 remobilization from the diagenetic mineralization. In the northwest, the Neoproterozoic (Dalradian
60 Supergroup) basement is similarly metamorphosed to greenschist facies, and contains chalcopyrite
61 in gold mines at Cononish (Scotland) and Curraghinalt (Northern Ireland), and in a former gold
62 prospect at Stronchullin (Patrick et al. 1988; Gunn et al. 1996; Parnell et al. 2000). These are
63 orogenic gold deposits, focussed on vein structures. At Cononish, sulphur isotope data indicate that
64 the fluid that deposited most of the gold and sulphides could have been sourced from other
65 metasedimentary units in the succession or was possibly of magmatic origin Spence-Jones et al.
66 (2018). The age of the mineralization is the same as the last stage of crystallization in nearby
67 granites, and the gold-bearing Rhynie Chert (Rice et al. 2012).

68 Paragenetic sequences for each sample type (Fig. 2) show that chalcopyrite is typically hosted in
69 quartz in the Neoproterozoic and Munster Basin deposits, and in dolomite or other carbonates in the
70 MVT deposits, and samples were selected from such veins (Table 1). Chalcopyrite was sampled from
71 specimens with a simple mineralogy, so that it was not contaminated by other phases (Fig. 3). Gold
72 in the Neoproterozoic, which occurs as tellurides or electrum, is coeval with chalcopyrite (Fig. 2).
73 Chalcopyrite is coeval with other sulphides in the other deposit types, but was sampled where it is
74 monomineralic.

75 **Methodology**

76 The cold crush method involves analysis by mass spectrometry conducted in high vacuum (Blamey
77 2012; Blamey et al. 2015). Each session was preceded and followed by analysis of 1 microlitre
78 capillary tubes for calibration. Atmosphere was also introduced to verify the calibration using 100 to
79 200 acquisitions for both the sample and atmosphere standard. A match head sized sample (about
80 250 microns) is crushed incrementally under a vacuum of $\sim 10^{-8}$ Torr, producing 4 to 10 successive
81 bursts, which remained in the vacuum chamber for 8-10 analyser scans (~ 2 s) before removal by the
82 vacuum pump. This method does not require a carrier gas and volatiles are not separated from each

83 other but released simultaneously into the chamber. The act of incremental crushing may open a
84 single inclusion or multiple fluid inclusions. The data acquisition is performed by means of two
85 Pfeiffer Prisma quadrupole mass spectrometers operating in fast-scan, peak-hopping mode.
86 Routinely the system analyses for the following gaseous species including H₂, He, CH₄, H₂O, N₂, O₂,
87 Ar, and CO₂. The volatiles are reported in mol%. The instrument is calibrated using Scott Gas Mini-
88 mix gas mixtures (with 2% uncertainty), capillary tubes filled with gas mixtures (with 1%
89 uncertainty), and three in-house fluid inclusion gas standards. The 2-sigma detection limit for most
90 inorganic species is about 0.2 ppm for aqueous fluid inclusions. Instrumental blanks were also
91 analyzed routinely to assess if gases were produced during the crushing process. The mass spectra
92 remained at background during crushing of blanks indicating that gases released are not sourced
93 from the crushers or hardware. The amount of each species was calculated by matrix methods to
94 provide a quantitative analysis, which is corrected for the instrumental background. Nine capillary
95 tubes with encapsulated atmosphere were analyzed and yielded N₂/Ar ratios of 83.2 with a standard
96 deviation of 1.4, within error of the atmospheric N₂/Ar ratio of 83.6. This translates into 0.5%
97 accuracy for artificial inclusions made under laboratory conditions. Precision using natural inclusions
98 for the major gas species measured is generally 2-5%, these being dependent on summed errors
99 derived from instrument noise, linearity of the mass spectrometer, uncertainty of standards, blanks,
100 interferences, and measurement of sensitivity factors. Before analysis, the crushing area and the
101 bellows of the crusher were cleaned using potassium hydroxide. The apparatus is also routinely
102 cleaned with isopropanol. Thereafter, the crushing chamber is baked at about 150-200 °C for 72 h
103 before loading and analysing the samples at room temperature the next day.

104 Results

105 Data were successfully measured in 27-25 sulphide samples (Fig. 1, Table 1). The gases in the
106 Carboniferous set is consistently almost exclusively (>99 %) water, and this contain very low
107 contents of carbon dioxide or methane. The data show no relationship to the lapetus suture. The
108 two greenschist-hosted sequences yield gas with a higher non-aqueous content (Fig. 24), which is
109 dominated by carbon dioxide (Table 1). The samples from the three gold-bearing localities in the
110 Neoproterozoic additionally contain variable contents of methane. In contrast, the samples from the
111 Munster Basin do not contain significant methane (Fig. 24). The total gas yield for each sample is a
112 weighted sum of incremental crushes: an example of the chemistry of successive crushes is given in
113 Fig. 35.

114 Discussion

115 Fluid types

116 The Carboniferous-hosted samples yield almost exclusively water, typical of inclusion fluids in
117 sedimentary basins. This includes the samples at Tynagh and Gortdrum, where the host sediments
118 have vitrinite reflectance values of about 4 % (Clayton et al. 1989), indicative of the gas window of
119 hydrocarbon generation. The host rocks in the Munster Basin to the south have higher reflectance
120 values of >5 % (Clayton et al. 1989) and are categorized as greenschist facies. The higher non-
121 aqueous component in both of the greenschist-hosted sample sets is typical of metamorphosed
122 sediments, in which carbon dioxide in particular becomes prevalent. The data for successive crushes
123 in a single sample show that the relative proportions of gas are relatively constant (Fig. 35). This is a
124 pattern characteristic of a single assemblage of fluid inclusions, as multiple assemblages (such as a

125 combination of primary and secondary inclusions) tend to exhibit progressive changes in chemistry
126 as the assemblages split open with different ease (Blamey 2012). Thus the data indicate that carbon
127 dioxide and methane occur together (Giggenbach 1980).

128 The inclusion of non-aqueous gases in both greenschist-hosted sequences is consistent with the
129 observation of carbonic fluid inclusions in all three gold-bearing deposits in the Neoproterozoic
130 (Patrick et al. 1988; Baron & Parnell 2005; Parnell et al 2000) and in the Munster Basin (Wen et al.
131 1996). However, while carbon dioxide was inferred from thermometric data, the additional presence
132 of methane was not recognised. The methane entrapped in the Neoproterozoic-hosted localities is
133 newly recognized here.

134 *Organic species*

135 Each of the three sample sets occurs in a sequence containing black shales. However in the Munster
136 Basin, the shales are limited and have not yielded carbon contents of oil-generating capacity (> 1%),
137 whereas the Carboniferous further north (Cornford 1998) and the Neoproterozoic (Bata & Parnell
138 2014) have both sourced large volumes of hydrocarbons [and so could yield methane](#). This may
139 explain the lack of methane in the Munster Basin. There is a difference in the occurrence of organic
140 species detected geochemically and petrographically. MVT deposits, to which the Carboniferous-
141 hosted deposits are attributed, commonly contain solid oil residues (bitumen), and such residues
142 occur in several of the deposits analysed in this study including [Laghy](#), Great Orme and Breedon.
143 However the Carboniferous-hosted deposits do not yield high methane values. Conversely, the gold-
144 bearing deposits contain methane, but do not contain solid organic residues, although the host
145 Neoproterozoic contains a large fossil oil reservoir in the same region (Bata & Parnell 2014).

146 These observations add weight to previous suggestions that methane plays a role in orogenic gold
147 mineralization (Naden & Shepherd 1989; Gaboury 2013; Goldfarb & Groves 2015; Jébrak et al.
148 2018). The data could be viewed as showing that the most methanic of the sample set are a guide to
149 gold-bearing deposits. The methodology therefore has potential in support of gold exploration.

150 *Other studies*

151 Data for gases are available for a limited number of gold-bearing deposits elsewhere. In particular
152 gas data from minerals in Carlin-type gold deposits in Nevada have also shown them to contain
153 significant methane (Blamey & Norman 2000; Blamey et al. 2017). The data from Britain and Ireland
154 are, in the case of opaque sulphides, from minerals which do not yield conventional fluid inclusion
155 microthermometric data, and so extend the range of minerals that can be investigated for direct
156 measurement of fluid chemistry.

157 Previous measurements of hydrocarbon gases extracted from limestones (Carter & Cazalet 1984;
158 Ferguson 1988; Mulshaw 1996) in the Carboniferous of Ireland and England show that anomalous
159 concentrations of gases may be detected in the vicinity of sulphide ore deposits, but that
160 observation is not matched by anomalies in the sulphides. The anomalies in the limestones may
161 reflect hydrocarbon generation due to thermal alteration of the host rock around the hydrothermal
162 systems. In contrast, the previously published fluid inclusion data for the greenschist-hosted
163 deposits show that carbon-bearing species were components of the mineralizing fluid.

164 **Conclusions**

165 The measurement of gases released by the crushing of chalcopyrite samples from Britain and
166 Ireland, at ambient temperature, shows that the resultant data may contain evidence for the nature
167 of the mineralizing fluids. In particular:

- 168 (i) The samples not heated to greenschist facies almost exclusively yield water.
- 169 (ii) The greenschist-hosted samples yield non-aqueous species, complementing
170 observations of CO₂ in fluid inclusions in accompanying minerals.
- 171 (iii) The Neoproterozoic samples, associated with gold mineralization, consistently contain
172 anomalous quantities of methane.
- 173 (iv) The data from opaque sulphides extend the range of phases that can be used to
174 characterize the mineralizing fluid.

175 These data add to previous studies of gas content in minerals from ore deposits, which have been
176 used for example to infer models for MVT deposits (Blamey 2012) and unconformity-related
177 uranium deposits (Rabiei et al. 2017), and the role of magmatic fluids (Blamey et al. 2017). Such
178 measurements contribute to an understanding of the origin of ore deposits, and thus can be
179 valuable in discriminating potential ore prospects during exploration.

180 **Acknowledgements**

181 We are grateful to J. Armstrong and C. Rice for provision of samples. J. Johnston is thanked for skilled
182 technical support.

183

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272

273 **Figure captions**

274 Fig. 1. Distribution of chalcopyrite samples in Britain and Ireland analysed for entrapped gases.
275 Sample localities: AL, Allihies; BC, Ballycummisk; BD, Breedon; BG, Ballygrant; BR, Brennand; CB,
276 Caldbeck; CL, Conlig; CN, Cononish; CT, Cloontoo; CT, Curraghinalt; GD, Gortdrum; GO, Great Orme's
277 Head; ~~LY, Laghy~~; NH, Nenthead; ~~NV, Navan~~; SC, Stronchullin; TY, Tynagh; WH, Wanlockhead.

278 Fig. 2. Paragenetic sequences for representative deposits of each type, at Conlig (MVT), Curraghinalt
279 (Neoproterozoic) and Ballycummisk (Munster Basin). Data modified from Moles & Nawaz (1996),
280 Parnell et al. (2000) and Wen et al. (1999).

281 Fig. 3. Photomicrographs of selected samples. A, Chalcopyrite (bright) in quartz, Cononish; B,
282 Chalcopyrite (bright) and specular haematite (grey) in quartz, Ballycummisk; C, Chalcopyrite (bright)
283 in dolomite, Tynagh.

284 Fig. ~~24~~. Cross-plot of non-aqueous gas content and methane content for sulphide samples. Two sets
285 of greenschist samples are distinguished by high and low methane contents.

286 Fig. ~~35~~. Chemistry of successive incremental crushes of a single sample of chalcopyrite from
287 Cononish. Similarity of increments shows that the chemistry reflects a single assemblage of fluid
288 inclusions.

289 Table 1. Compositional data for gases released from sulphide samples, Britain and Ireland

Lab no.	Locality	Mineral	Type	H ₂ O (%)	non-H ₂ O (%)	CH ₄ (%)	CO ₂ (%)	Total mols	%CH ₄ (non-H ₂ O)	%CO ₂ (non-H ₂ O)
296	Cononish	Chalcopyrite in quartz	Neoproterozoic psammite	98.36	1.64	0.68	0.97	2.52E-09	0.41	0.59
294	Cononish	Chalcopyrite in quartz	Neoproterozoic psammite	99.44	0.56	0.22	0.34	3.49E-09	0.39	0.61
320	Cononish	Chalcopyrite in quartz	Neoproterozoic psammite	87.43	12.57	5.09	7.46	4.20E-09	0.40	0.59
304	Stronchullin	Chalcopyrite in quartz	Neoproterozoic psammite	95.68	4.32	0.07	4.18	1.15E-10	0.02	0.97
305	Stronchullin	Chalcopyrite in quartz	Neoproterozoic psammite	93.78	6.22	0.1	6.1	9.44E-10	0.02	0.98
315	Curraghinalt	Chalcopyrite in quartz	Neoproterozoic psammite	55.9	44.1	10.44	33.54	1.53E-09	0.24	0.76
316	Curraghinalt	Chalcopyrite in quartz	Neoproterozoic psammite	65.93	34.07	8.16	25.79	4.87E-09	0.24	0.76
314	Ballygrant	Chalcopyrite in dolomite	MVT in Neoproterozoic Lst	99.73	0.27	0.02	0.13	1.81E-11	0.07	0.48
302	Great Ormes Head	Chalcopyrite in dolomite	MVT in Carboniferous Lst	99.91	0.09	0.004	0.016	1.80E-11	0.04	0.18
303	Brennand	Chalcopyrite in calcite	MVT in Carboniferous Sst	99.92	0.08	0.003	0.04	4.71E-12	0.04	0.50
309	Langness	Chalcopyrite in calcite	MVT in Carboniferous Sst	99.43	0.57	0.03	0.5	2.44E-10	0.05	0.88
321	Nenthead	Chalcopyrite in siderite-ankerite	MVT in Carboniferous Lst	99.7	0.3	0.06	0.03	4.18E-11	0.20	0.10
323	Breedon	Chalcopyrite in dolomite	MVT in Carboniferous Lst	99.87	0.13	0.01	0.08	1.61E-11	0.08	0.62
9799	Laghy	Sphalerite	MVT	99.66	0.34	0.23	0.08	1.95E-09	0.68	0.24

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306	Leadhills	Chalcopyrite <u>in dolomite-calcite</u>	MVT <u>in Ordovician pelite</u>	98.35	1.65	0.02	1.58	1.75E-10	0.01	0.96
307	Leadhills	Chalcopyrite <u>in dolomite-calcite</u>	MVT <u>in Ordovician pelite</u>	99.95	0.05	0.01	0.01	7.09E-12	0.20	0.20
312	Caldbeck Conlig	Chalcopyrite <u>in dolomite</u>	MVT <u>in Ordovician volcanics</u>	99.96	0.04	0.01	0.01	2.38E-11	0.25	0.25
318	Conlig	Chalcopyrite <u>in dolomite-barite</u>	MVT <u>in Silurian pelites</u>	99.27	0.73	0.01	0.22	2.89E-10	0.01	0.30
319	Conlig	Chalcopyrite <u>in dolomite-barite</u>	MVT <u>in Silurian pelites</u>	99.51	0.49	0.014	0.2	3.95E-10	0.03	0.41
327	Navan	Galena	MVT	98.45	1.55	0.47	0.9	6.39E-10	0.30	0.58
300	Tynagh	Chalcopyrite <u>in dolomite</u>	MVT <u>in Carboniferous Lst</u>	99.05	0.95	0.01	0.86	1.61E-09	0.01	0.91
301	Tynagh	Chalcopyrite <u>in dolomite</u>	MVT <u>in Carboniferous Lst</u>	99.26	0.74	0.01	0.65	1.35E-09	0.01	0.88
293	Gortdrum	Chalcopyrite <u>in dolomite-quartz</u>	MVT <u>in Carboniferous Lst</u>	99.9	0.1	0.001	0.04	3.68E-10	0.01	0.40
					100				0.00	0.00
297	Allihies	Chalcopyrite <u>in quartz</u>	Munster Basin <u>Devonian Sst</u>	86.86	13.14	0.01	12.64	2.79E-09	0.00	0.96
325	Allihies	Chalcopyrite <u>in quartz</u>	Munster Basin <u>Devonian Sst</u>	98.59	1.41	0.002	0.43	6.76E-12	0.00	0.30
299	Ballycummisk	Chalcopyrite <u>in quartz-dolomite</u>	Munster Basin <u>Devonian Sst</u>	89.93	10.07	0.004	6.26	3.06E-10	0.00	0.62
311	Ballycummisk	Chalcopyrite <u>in quartz-dolomite</u>	Munster Basin <u>Devonian Sst</u>	94.57	5.43	0.0002	0.88	3.12E-11	0.00	0.16
317	Cloontoo	Chalcopyrite <u>in quartz</u>	Munster Basin <u>Devonian Sst</u>	95.51	4.49	0.06	4.18	4.74E-11	0.01	0.93

290

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292