- 1 Mutual interaction of phenolic compounds and microbiota: Metabolism of
- 2 complex phenolic apigenin C- and kaempferol O-derivatives by human fecal
- 3 **samples**

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16 **ABSTRACT**

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Human colonic bacteria have an important impact on the biotransformation of flavonoid glycosides and their conversion can result in the formation of bioactive compounds. However, information about the microbial conversion of complex glycosylated flavonoids and the impact on the gut microbiota are still limited. In this study, in vitro fermentations with selected flavonoid O- and C-glycosides and three different fecal samples were performed. As a result, all flavonoid glycosides were metabolized via their aglycones vielding smaller substances. Main metabolites were 3-(4-hydroxyphenyl)propionic acid, 3-phenylpropionic acid, and phenylacetic acid. Differences in the metabolite formation due to different time courses between the donors were determined. Therefore, from all fermentations, the ones with a specific donor were always slower resulting in a lower number of metabolites compared to the others. Exemplarily, tiliroside was totally degraded from 0h (105 \pm 13.2 μ M) within the first 24h, while in the fermentations with fecal samples from other donors, tiliroside $(107 \pm 52.7 \,\mu\text{M} \text{ at 0h})$ was not detected after 7h anymore. In general, fermentation rates of C-glycosides were slower compared to the fermentation rates of Oglycosides. The O-glycoside tiliroside was degraded within 4h while the gut microbiota converted the C-glycoside vitexin within 13h. However, significant changes (p < 0.05) in the microbiota composition and short chain fatty acid levels as products of carbohydrate fermentation were not detected between incubations with different phenolic compounds. Therefore, microbiota diversity was not affected and a significant prebiotic effect of phenolic compounds cannot be assigned to flavonoid glycosides in food-relevant concentrations.

41 **KEYWORDS**

- Flavonoid glycosides, metabolism, in vitro fermentation, human gut microbiota, short
- 43 chain fatty acids

45 **INTRODUCTION**

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the plant kingdom. Their chemical structure is characterized by two aromatic rings with at least one hydroxyl group. 1 Flavonoids can be classified into several subgroups depending on the constitution of the heterocyclic C-ring.² The chemical structures range from very simple substances to guite complex flavonoids consisting of several phenolic rings with a wide range of substituents. In plants, they are frequently conjugated with different small organic molecules (preferentially sugars and organic acids), affecting their water-solubility and functionality.³ Only occasionally they are present in plants as non-conjugated flavonoid aglycones.¹ Beside the very frequent O-glycosylation of flavonoids, flavonoid glycosides can also occur as C-glycosides.4 The most common sources of flavonoid C-glycosides are swiss chard, tomato, lemons, and some cereals such as maize, wheat, and rice.^{5, 6} When consumed with plant foods, flavonoid glycosides are hypothesized to have health beneficial properties.⁷ Antimutagenic, anticarcinogenic, antiviral, antibacterial, and antiinflammatory properties have been described.8-11 Even though studies have mainly focused on the more common O-glycosides, it is assumed that C-glycosides would have better therapeutic properties due to their enhanced stability over the respective aglycone and O-glycosides. 12 But especially for C-glycosides, more investigations on resorption, metabolism, and health beneficial effects have to be carried out.¹² However, uptake/bioavailability of flavonoids is a crucial factor for potential bioactivities but this is still discussed controversially. 13 It is highly dependent on the chemical structure and can therefore differ between different compounds. Moreover, bioactivity is influenced by the resorption rate, intestinal, and hepatic metabolism, and the subsequent distribution in the organism. 14 Although different

Flavonoids are a class of secondary plant metabolites which are found ubiquitously in

pathways have been identified as being responsible for an enhanced resorption, it has to be concluded that in most cases resorption rates of intact flavonoids are not very high.^{15, 16} Consequently, the most important metabolic transformation of flavonoid glycosides takes place in the colon, where they undergo significant degradation by the gut microbiota.

In the human colon more than 500 different bacterial species are found, whereby the composition of the microbiome is highly diverse and unique for every individual. ^{7, 17} Some of the bacterial species can catalyze (flavonoid) *O*- and *C*-deglycosylation, demethylation, dehydroxylation, ester cleavage, reduction of carbon-carbon double bond, isomerization, ring fission, and decarboxylation. ¹⁸ However, not all bacteria are able to carry out every degradation step. ¹⁹ For example, *Enterococcus casseliflavus* is only able to cleave and ferment the sugar moiety from different quercetin-*O*-glycosides to short chain fatty acids (SCFA), while the aglycone is not degraded any further. In contrast, *Eubacterium ramulus* is capable of converting also the aglycone. ²⁰ Additionally, it is possible that some kind of cross-feeding effects due to the degradation of phenolic compounds occur. ²¹ It means that one strain gives good growth on the primary substrate such as the phenolic compound, which is metabolized to a product being a secondary substrate for another strain. This might enhance growth of further microorganisms and consequently influence composition of the colonic microbiota. ²²

Although the structures of *O*- and *C*-glycosides are very different from a chemical point of view, the conversion of *C*-glucosides also showed a dependence on the presence of specific bacteria for degradation steps such as the cleavage of the C-C bond. For example, *Eubacterium cellulosolvens* is not able to deglycosylate the *C*-

glycosides vitexin, while it is degradable by the intestinal *Lachnospiraceae* strain CG19-1.²³

When looking at fermentation experiments with fecal samples, it is possible to investigate at least two ways of action: On the one hand, the microbial transformation of flavonoids to bioactive products is of interest, whereas on the other hand, the influence of phenolic compounds and their metabolites on the composition of the gut microbiota is an important aspect, too (two-way phenolic - microbiota interaction).²⁴ In general, combining metagenomics and metabolomics studies will help to better understand both types of interactions.²⁵ However, information about the microbial transformation of more complex, highly glycosylated flavonoids, and evidence for their effect on the gut microbiota is still not sufficiently investigated. In some studies, it has been suggested that more complex polyphenols have an even higher effect on the microbiome than simple structures, because a larger variety of potentially bioactive metabolites can be formed.²⁶ Therefore, such structures might also influence the diversity of the microbiota which is hypothesized to correlate with the health of an individual.²⁷

The aim of this study was to investigate the degradation of selected structurally related, highly glycosylated flavonoids and their influence on gut microbiota composition. Special focus was set on the comparative assessment of group specific differences between *O*- and *C*-glycosides. For this purpose, different *in vitro* fermentations with a limited number of human fecal samples were performed. Breakdown products were analyzed using HPLC-ESI-MS/MS analysis. To assess whether phenolic exposure influenced the microbiota, qPCR and Illumina sequencing were carried out to investigate potential compositional changes and SCFA production was determined to detect major functional changes.

MATERIAL AND METHODS

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Chemicals. Apigenin (AP, 4',5,7-trihydroxyflavone), kaempferol (K, 3,4',5,7-120 tetrahydroxyflavone), kaempferol-3,4'-O-diglucoside-7-O-rhamnoside (K-DG-R), 121 tiliroside (T, kaempferol-3-(6"-trans-p-coumaroyl)glucoside), vitexin (V, apigenin-8-C-122 glucoside), and vitexin-2"-O-rhamnoside (V-R) were purchased from Phytolab GmbH 123 & Co. KG (Vestenbergsgreuth, Germany). Vitexin-2"-O-glucoside (V-G), and vitexin-124 2"-O-xyloside (V-X) were isolated from swiss chard. The isolation method is 125 described in the supplementary information. 126 127 The chemical structures of the substrates used in this study are shown in Figure 1. All O-glycosides are based on the aglycone structure of kaempferol, whereas the 128 structures of the C-glycosides are related to the aglycone apigenin. Both aglycones 129 130 differ only in a hydroxyl group at C3-position of the basic skeleton. While tiliroside has only one sugar moiety being bound via a C-O bond at the C3-position and further 131 esterification with p-coumaric acid, kaempferol-3,4'-O-diglucoside-7-O-rhamnoside is 132 O-glycosylated with three sugar moieties at different positions. In contrast, vitexin is 133 the simplest C-glucoside of apigenin, bound via a C-C bond at the C8-position. The 134 different vitexin derivatives have a second sugar moiety at the C2"-position. These 135 bonds are C-O bonds. Consequently, the vitexin derivatives combine O- and C-136 137 glycosidic bonds in one structure. In vitro degradation of phenolic substrates by fecal microbiota. This study was 138 part of a set of experiments already described by Vollmer, et al. 28 in which three 139 independent in vitro fermentations with fecal microbiota were performed. Fecal 140 samples were donated by three different, healthy donors (donor A, B, and C) without 141 a history of gastrointestinal disorders or any antibiotics consumption for at least three 142 month prior to the fermentation experiment. Number of samples and fermentation 143

strategy were similar to related fermentation experiments already described in the literature. $^{21, 29\cdot31}$ The preparation of the fecal suspension was conducted according to Vollmer, Schröter, Esders, Farquharson, Neugart, Duncan, Schreiner, Louis, Maul and Rohn 28 . *In vitro* fermentations were performed with eight different flavonoids in triplicate at an initial pH value of 6.5 in a final volume of 10 mL. Based on preliminary experiments with similar levels of carbon source(s) it was estimated that during fermentation the pH drops by 0.5-1 units. The volume of 10 mL consisted of 9.4 mL fermentation medium which contained several minerals, supplements, and carbohydrates to allow the gut bacteria to grow, 100 μ L phenolic substrate (final concentration 200 μ M, pre-dissolved in DMSO), 14 μ L vitamin solution, and 500 μ L freshly prepared fecal suspension (0.2% final fecal concentration). The preparation and ingredients of the fermentation media and vitamin solution are described elsewhere. 28

After adding the fecal suspension to the medium, the samples were incubated on a rotator (Stuart SB3, Bibby Scientific, Stone, UK) for 48 h at 37 °C and 25 rpm. Two aliquots of 750 µL were collected initially (0h), and after 3 h, 7 h, 24 h, and 48 h while flushing with CO₂. One aliquot was frozen with liquid nitrogen and stored at -80 °C until HPLC-ESI-MS/MS analysis. The second aliquot was used for investigations of the microbiota and SCFA analyses. For that, the aliquot was separated into supernatant (SCFA analysis) and cell pellet (DNA analysis) by centrifuging.²⁸ Supernatant was stored at -20 °C until SCFA analysis and cell pellet dissolved in buffer at -80 °C. Incubation parameters of fecal fermentation and sample treatments were already described by Vollmer, Schröter, Esders, Farquharson, Neugart, Duncan, Schreiner, Louis, Maul and Rohn ²⁸.

In addition to fecal suspensions with phenolic substrates, control samples for every fermentation without any substrates were conducted. This was done to determine the metabolite formation caused by compounds endogenously present in the fecal samples. For that, 100 μ L DMSO were used instead of the substrate ('control samples I'). For the estimation of the interactions of the substrates with the medium, 100 μ L substrate were combined with 9.9 mL medium and 14 μ L vitamin solution and were treated the same way as the samples ('control samples II').

The fermentation period in the present study was set to 48 h based on other *in vitro* fermentation experiments on phenolic compounds described in the literature. ^{19, 32-34} Some of the fermentation experiments ended already after 24 h, but there, mainly simple phenolic substances were investigated. Due to the assumption that the microorganisms need longer for the degradation process of complex phenolic compounds the fermentation period was extended to 48 h.

Sample preparation and HPLC-ESI-MS/MS analysis for quantification. The substrates and their metabolites were extracted with ethyl acetate from 400 µL of the fermentation mixture from every time point. The sample preparation and extraction were carried out according to Vollmer, Schröter, Esders, Farquharson, Neugart, Duncan, Schreiner, Louis, Maul and Rohn ²⁸.

HPLC analyses of the substrates and their metabolites were performed using a 1260 Infinity Series system from Agilent Technologies Deutschland GmbH & Co. KG (Waldbronn, Germany). The system consisted of a binary pump, an online-degasser, an autosampler, and a column oven. The separation of the substances and their metabolites was carried out using a Kinetex 5 u EVO C18 100 A (150 x 2.1 mm) column equipped with an EVO C18 pre-column (both from Phenomenex Inc., Aschaffenburg, Germany). The chromatographic separation took place at 20 °C. The

mobile phase was 0.1% formic acid in water (A) and 0.1% formic acid in acetonitrile (B) and the following gradient elution was used: 5% B (0-1 min), 11% B (1-2 min), 14% B (2-4 min), 50% B (4-15 min), 95% B (15-18 min), 95% B (18-22 min), 5% B (22-23 min) and 5% B (23-32 min). The flow rate was set to 300 μ L/min and the injection volume was 5 μ L.

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For detecting the substrates and their metabolites, mass analyses were performed in negative ion mode on an API2000 triple quadrupole MS/MS system (AB Sciex Germany GmbH, Darmstadt, Germany) equipped with an ESI interface. The following mass spectrometer settings were used: electrospray voltage = -4500 V, temperature = 450 °C, curtain gas = 1.7 bar, ion source gas 1 = 2.1 bar, ion source gas 2 = 5.2 bar and collisions gas = 0.5 bar (all nitrogen). For each substance, the optimum settings of the declustering potential, entrance potential, collision energy, collision exit potential, and the characteristic fragments were also determined.

For quantification, an external matrix calibration (0-200 µM) for every donor was used (**Supplementary Table 1**). The matrix consisted of the fermentation medium and the fecal sample. The data obtained were analyzed with the software Analyst® 1.5.2 from AB Sciex Germany GmbH (Darmstadt, Germany).

Beside the substrates used, the following metabolites were included in the method: benzoic acid, 4-hydroxybenzoic acid, caffeic acid, p-coumaric acid (p-CA), 3-(3,4dihydroxyphenyl)propionic acid, 3-(3,4-dihydroxyphenyl)acetic acid, 3,4dihydroxytoluene, ferulic acid, 3-(3-hydroxyphenyl)propionic acid, 3-(4hydroxyphenyl)propionic acid (4-HPPA), 3-phenylpropionic acid (3-PPA), 3hydroxyphenylacetic acid (PAA), 4-hydroxyphenylacetic acid, phenylacetic acid (PAA), hippuric acid, and kaempferol-3-O-glucoside (K-G).

Analysis of short chain fatty acids. Determination of short chain fatty acids (SCFA) including acetate, propionate, butyrate, iso-butyrate, formate, but also of further organic acids such as lactate, succinate, valerate, and iso-valerate was conducted with 500 μL of the supernatants from the 48 h samples and from one 0 h sample per substrate. The sample preparation was based on the method developed by Richardson, et al. ³⁵ followed by gas chromatography analysis using a Hewlett-Packard gas chromatograph fitted with a fused silica capillary column with helium as a carrier gas. Calculations were done with external standards, 2-ethylbutyrate was used as internal standard. The mean of the SCFA concentration resulting from the fermentations with the different substrates were set in relation to the SCFA concentration from the 'control samples I'.

DNA extractions, fluorimetric DNA quantification and qPCR analysis. DNA extractions were done with 400 μL of the fecal suspension from every donor and with the cell pellets from the 48 h samples using the FastDNA® spin kit for soil (MP Biomedicals, Illkirch, France). Extraction, fluorimetric DNA quantification and qPCR analysis were performed according to the protocol described by Vollmer, Schröter, Esders, Farquharson, Neugart, Duncan, Schreiner, Louis, Maul and Rohn ²⁸. Universal primers against total bacteria and specific primers against *Bifidobacterium* spp., Bacteroidetes, Ruminococcaceae, Lachnospiraceae, *F. prausnitzii*, *Blautia* spp., and the *Roseburia/Eubacterium rectale* group, were used.²⁸

Illumina sequencing of 16S rRNA genes. The Illumina sequencing of selected compounds was performed at the ZIEL – Institute for Food & Health of the TU Munich with DNA extracts of the fecal suspension from every donor (initial composition), the 48 h sample of tiliroside from all three donors, and the 48 h samples of vitexin, vitexin-2"-O-rhamnoside and kaempferol-3,4'-O-diglucoside-7-O-

rhamnoside from donor A and C. The method details are described in the supplementary information. 429,958 initial sequence reads were filtered as described in supplementary information, which resulted in 245,816 final reads (11,698-19,887 per sample). Those were assigned to 151 operational taxonomic units (OTUs) at ≥97% sequence identity (**Supplementary Table 2**). Sequencing data generated during this study are available in the SRA database under SRA accession SRP117249 at http://www.ncbi.nlm.nih.gov/sra/SRP117249.

Statistical analysis. Statistical analysis was performed to investigate whether differences in SCFA levels or in the microbiota composition resulting from the substrate fermentations were significant or not compared to the 'control samples I'. For that, data were analyzed using IBM SPSS Statistics 22 (Ehningen, Germany). All data were tested for normality. With the normal distributed data investigations for significant differences were carried out with an independent t-test. Non-normal distributed values were analyzed with the non-parametric Mann-Whitney-U-test. For significance, a confidence level of 95% (p < 0.05) was used. In addition, the qPCR, SCFA, and sequence data were summarized by Principal Component Analysis (PCA). All data were standardized before applying PCA. The analysis was performed in R (R Core Team (2012). R: A language and environment for statistical computing. R Foundation for Statistical Computing, Vienna, Austria. http://www.R-project.org). For correlation analysis between qPCR and sequencing data, OTUs were assigned to the corresponding qPCR assays based on their taxonomy (Supplementary Table 2).

RESULTS AND DISCUSSION

In the present study, the degradation of selected, structurally related, highly glycosylated flavonoids were determined in three independent *in vitro* fermentations in order to estimate metabolite formation and changes in the microbiota composition. Here, it was of interest to determine if the gut microbiota is able to convert the more complex flavonoids to lower molecular metabolites, or if they are not or only partially degraded. For comparison and as a control, *in vitro* fermentations with the corresponding aglycones were also performed.

In vitro fermentations of the flavonol kaempferol and its *O*-glycosides tiliroside and kaempferol-3,4'-*O*-diglucoside-7-*O*-rhamnoside.

Kaempferol. When starting the fermentation experiments with the comparatively simple aglycone structure kaempferol (K), there was only a slow degradation by the fecal sample from donor A. K was still present at 48 h and only small amounts of the metabolite 3-(4-hydroxyphenyl)propionic acid (4-HPPA) were detected (Table 1). However, K was fully metabolized by 7 h and 24 h, respectively, with the fecal sample from donors B and C and 4-HPPA was detected at 7 h at its highest concentration in both cases. During the fermentation with the fecal sample from donor B no further metabolites were present after 24 h. By contrast, the metabolite phenylacetic acid (PAA) was detected in the fermentation with the fecal sample from donor C. In the HPLC-ESI-MS/MS method used, the detector sensitivity for the metabolite 3-phenylpropionic acid (3-PPA) and PAA was rather low. It cannot be judged whether e.g., PAA was only formed to lower amounts by donor B unlike for the other two donors, because responsible microorganisms are missing or the amount was simply lower than the limit of detection (LOD) of the method. Therefore, it is possible that the formation of 3-PPA and PAA could not be detected in the

fermentation B. Nevertheless, the results obtained are in agreement with results described in the literature, where mainly 4-hydroxyphenylacetic acid, the precursor for PAA, which was also detected in the present study, has been described as an important phenolic metabolite (**Figure 2**).^{7, 36}

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Tiliroside. In all three in vitro fermentation experiments, tiliroside (T) was metabolized significantly by the gut microbiota. Figure 3 displays its degradation steps and metabolite formation. The detected metabolites were kaempferol-3-O-glucoside (K-G), kaempferol (K), p-coumaric acid (p-CA), 3-(4-hydroxyphenyl)propionic acid (4-HPPA), and 3-phenylpropionic acid (3-PPA), whereby the latter was not detected in the fermentation with the fecal sample from donor A. In the fermentations of donor B and C, the bacteria already cleaved the ester bond between the sugar moiety and p-CA within short time resulting in the formation of K-G at 0 h, immediately after inoculation with the fecal samples (Table 1, Figure 3). An interaction or breakdown due to the fermentation media can be excluded, because K-G was not detected in the 'control samples II' (substrate + medium, data not shown). The detection of K-G leads to the assumption that the microbiota did not mainly first cleave the bond between the kaempferol-moiety and the sugar substituent in T, and T is mostly degraded to its aglycone K via the intermediate product K-G. However, the metabolite K-G was only detected in very low concentrations in fermentation A at 0 h, 3 h, and 7 h. Instead, a significant increase of K and p-CA was present within the first 7 h. This may be due to the cleavage of the bond between K and glucose and the release of p-CA taking place at the same time intervals. Subsequently, the aglycone was transformed to smaller phenolic acids deriving from the A- and B-ring during the breakdown of the heterocyclic flavonoid C-ring.²⁵ Furthermore, the microbiota seems also to be able to use the free p-CA as a substrateas already described in some

studies where different strains were able to convert p-CA. As a result, further small metabolites such as 4-HPPA are formed (**Figure 2**).^{37, 38} Therefore, the formation of 4-HPPA derives from the degradation of the aglycone structure and p-CA, being a structural part in the chemical structure T.

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While K was metabolized only slowly in the fermentation of T with the fecal sample from donor A and was still present in a very high concentration at 48 h, p-CA was degraded continuously up to 24 h. This resulted in the formation of 4-HPPA corresponding to the reduction of the double bond, which is a very typical reaction in the microbial transformation of cinnamic acid derivatives.³⁹ At 48 h, 4-HPPA showed a similar concentration compared to the 24 h samples indicating that no significant further degradation of 4-HPPA was carried out by the microbiota from the fecal sample of donor A. In the fermentations with the fecal samples from donors B and C, 4-HPPA was already detected at 7 h at its highest concentration. Subsequently, 4-HPPA was fully metabolized to 3-PPA by a dehydroxylation at the C4-position. This is in accordance with a study published by Scheline 40, who found that the reduction of the double bond by fecal microorganisms greatly exceeds dehydroxylation reactions. Differences between the fermentations B and C can be found in the concentration of the intermediary metabolites p-CA and 4-HPPA. Both had their concentration maximum at 7 h. While the concentration of p-CA in fermentation B was much lower than the concentration of 4-HPPA, the concentration of both metabolites was almost identical in fermentation C. It is well known, that many anaerobic bacteria are not able to carry out every degradation step and therefore metabolize aromatic compounds not completely. 19 Consequently, it might be possible that the microbiota composition of donor B was better adapted for carrying out reduction reactions than the microbiota of donor C.

Taking the results of the *in vitro* degradation of the reference compound K into account (**Table 1**), only 4-HPPA and PAA were detected. Therefore, it can be assumed that *p*-CA in the fermentations with T was mainly detected because of being a structural part of T. As already mentioned, *p*-CA is also the precursor for the formation of 4-HPPA (**Figure 2**). Therefore, the amount of 4-HPPA in all fermentations with T as substrate was much higher than the concentration of 4-HPPA in the fermentations with the substrate K alone.

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Kaempferol-3,4'-O-diglucoside-7-O-rhamnoside. Compared to the fermentation of T, where several metabolites were identified, a lower number of phenolic metabolites was determined in the fermentation of the more complex kaempferol-3,4'-Odiglucoside-7-O-rhamnoside (K-DG-R, Table 1). Due to the fact that K-DG-R itself was not stable in the ion source and therefore, could not be detected with the analytical method used, the degradation was investigated only by the formation and degradation of K-G and K. In the 'control samples II', where the substrate was incubated with the medium, neither K-G nor K were detected (data not shown). Consequently, the formation of the different metabolites resulted only from the microbial degradation of K-DG-R. For quantification, only the metabolite K-G could be included in the method, due to the lack of reference compounds. However, additionally qTOF analyses were carried out and the tentative identification of the metabolites kaempferol-3,4'-O-diglucoside, kaempferol-3-O-glucoside-7-Orhamnoside, or kaempferol-4'-O-glucoside-7-O-rhamnoside, and kaempferol-7-Orhamnoside was possible. The metabolites were detected at the same fermentation times as K-G (data not shown). Furthermore, the metabolites K-G (0-7 h), K (3-48 h), and 4-HPPA (48 h) for the fermentation with the fecal sample from donor A and K-G (0-3 h), K (3 h), and PAA (24-48 h) in the fermentations with the fecal sample from donors B and C were detected and quantified. In conclusion, as already shown for the substrate T, K-DG-R was degraded to phenolic acids via the consecutive release of the sugar moieties and the aglycone.

In comparison to the reference compound K, similar metabolites were detected. But, in the fermentation B and C of K-DG-R only the phenolic acid PAA but not the metabolite 4-HPPA was present compared to the fermentation of K. It is possible that the formation of PAA went via 4-HPPA which was not detected in the fermentations B and C, probably resulting from the time intervals, when the samples were taken (**Table 1, Figure 2**).

In vitro fermentation of the flavone apigenin and its C-glycosides vitexin, vitexin-2"-O-rhamnoside, vitexin-2"-O-glucoside and vitexin-2"-O-xyloside. In the *in vitro* fermentations of apigenin (AP) and its C-glycosidic derivatives, all three human fecal microbiomes were able to convert the substrates yielding smaller phenolic acids. Analogously to the fermentation of the phenolic compounds mentioned above, differences between the donors in the metabolite formation were determined. Furthermore, the fermentations with the fecal sample from donors B and C showed the formation of smaller molecular metabolites than the fermentation with the fecal sample from donor A. Also, the same metabolites, 4-HPPA, PAA, and 3-PPA were detected, but the intermediate products differed between the initial substrate.

Apigenin. In the fermentation with the fecal sample from donor A, AP was still detected at 48 h, while in the fermentation with the fecal sample from donors B and C, there were only low concentrations of AP at 0 h. Consequently, as AP was not present in the 'control samples II' (data not shown), an interaction between the substrate and the medium can be excluded, and it appears that it was already

partially degraded. In general, the formed metabolites were 4-HPPA (A: 24 h, 48 h; B: 7 h, C: 3 h, 7 h) and PAA (A: 48 h; B: 48 h, C: 24 h, 48 h) for all three fermentations and additionally 3-PPA (B: 24 h, 48 h, C: 24 h, 48 h) for the fermentations with the fecal sample from donors B and C (**Table 2**).

Vitexin. The fermentations with the substrate vitexin (V) showed the same phenolic acid metabolites as the fermentation of AP, with different maxima: 4-HPPA (A: 24 h, 48 h; B: 24 h, 48 h), PAA (A: 48 h; B: 24 h, 48 h, C: 24 h, 48 h), and 3-PPA (B: 24 h, 48 h; C. 24 h, 48 h) (Table 2). V itself was degraded within the first 24 h in the fermentations with the fecal samples from donors B and C but only disappeared after 48 h in the fermentation with the fecal sample from donor A. Additionally, the metabolite AP was detected at 48 h in the fermentation with the fecal sample from donor A. Therefore, the degradation of V seems to take place via the formation of the aglycone structure and the glucose moiety was primarily cleaved by the bacteria. Due to faster metabolism, AP was probably not detected in the fermentations with the fecal samples from donors B and C. The metabolites observed are in agreement with descriptions in the literature, where for example the rod-shaped Gram-positive *Lachnospiraceae* strain CG19-1, was able to convert the substrate V.²³ In contrast, two *Lactococcus species* and one *Enterococcus species* were not able to convert V, as described by Kim, et al. ⁴¹.

<u>Vitexin derivatives</u>. In the fermentations with the different vitexin derivatives, the concentrations of vitexin-2"-O-glucoside (V-G), vitexin-2"-O-rhamnoside (V-R), and vitexin-2"-O-xyloside (V-X) were very low or initial substances not even detectable in the fermentation samples. In contrast to V, those compounds have a second sugar moiety bound which increases their water solubility. It is assumed that the vitexin derivatives still remain in the aqueous fermentation medium when extracting the

substrates and metabolites with ethyl acetate. Due to the fact, that the 'control samples II' (substrate + medium) did not show any phenolic breakdown products and substrate concentrations in the 'control samples II' and fermentation samples were very similar, the formed metabolites can be attributed to the degradation of the corresponding initial substances (data not shown). Consequently, the degradation of the vitexin derivatives could only be displayed by their metabolite formation.

In all *in vitro* fermentations with the different vitexin derivatives, the degradation went via the intermediate product V indicating that the first step was cleavage of the O-glycoside bond by the microbiota. On the basis of the formation of V, it was possible to compare the release of the second glycoside moiety. It can be seen that in fermentation B and C the release of the glucose moiety was quicker than the release of rhamnose, or xylose (**Table 2**). **Figure 4** shows exemplarily the V formation within the *in vitro* fermentation with the fecal sample from donor B for the three different vitexin derivatives investigated. Graphs were normalized to 100%. This effect may be due to the fact that glucoside units are more common in nature and more microorganisms are adapted to utilize glucose linked to secondary plant phenolics than other sugars. Despite the intermediate metabolite V, the main end products of the fermentation of the vitexin derivatives were 4-HPPA, PAA, and only 3-PPA in one case, which is in accordance with the metabolites resulting from the degradation of V (**Table 2**).

The aglycone AP was not detected in the fermentations of the different vitexin derivatives, but similar metabolites were detected compared to the fermentation of pure AP (**Table 2**). Due to the fact, that the metabolite V was present and V itself appears to be degraded to phenolic acids via the intermediate product AP, it was

assumed that the formation of the phenolic acids went via the aglycone structure as well. However, this was not detectable in the present study.

It is recognizable that the recovery rates of the initial substrates in some fermentations (i. e., V-G) were very low compared to the amount of the phenolic substrates in the fermentation mixture. When looking at the 'control samples II', where the phenolic compounds were co-incubated with the medium, it is notably that the concentrations of the phenolic compounds were not higher and very similar compared to the ones in the fermentation mixtures (data not shown). Therefore, one possible reason for the low concentrations is interactions between the compounds and the medium. It is also possible that the compounds precipitate or are adsorbed without metabolism by the microbial biomass, the solid components of the media or the inoculum. Furthermore, due to their different hydrophilicity, it is possible that more hydrophilic substrates still remain in the aqueous medium when extracting with ethyl acetate.

Before starting the fermentation experiments, different extraction procedures were investigated and optimized for an overall mixture of compounds in buffered aqueous media, similar to the one used in the present study, by looking at the recovery rates after the extraction (data not shown). Due to the fact that the substrates and metabolites used in this study show very different polarities and solubilities the methodological approach was a compromise for covering substrates (comparatively more hydrophobic) and metabolites (comparatively more hydrophilic).

qPCR analysis, Illumina sequencing and SCFA production. Overall microbiota composition and activity after 48 h of incubation were assessed by qPCR for different bacterial groups (total bacteria, *Bifidobacterium* spp., Bacteroidetes, Ruminococcaceae, Lachnospiraceae, *F. prausnitzii*, *Blautia* spp., and the

Roseburia/Eubacterium rectale group) and determination of short chain fatty acid production in order to investigate a possible effect of the phenolic substrates on the microbiota. Significant differences (p < 0.05) between the 'control samples I' (medium + fecal sample without any substrate) and the different fermentations after 48 h of incubation were not determined for any of the bacterial groups or SCFA (Supplementary Table 3). The PCA (Figure 5) based on the relative qPCR and SCFA results shows that the different samples, even after 48 h of incubation, clustered by donor and not by the different phenolic substrates used. Therefore, the different phenolic compounds did not have a major effect on the microbiota. Furthermore, it is assumed that the SCFA in this study were mainly formed from the carbohydrates present in the fermentation media and fermenting the sugar moiety of the flavonoid glycosides did not lead to a significant difference because of the quite low concentrations of these compounds in the fermentation mixture.

Additionally, Illumina sequencing was performed with selected samples to investigate whether specific bacteria were affected by the presence of the phenolic substrates that may not have been detected by the qPCR analysis. Correlation between qPCR data and corresponding sequence data overall showed good agreement (Supplementary Figure 1). Small variations are likely due to slight differences in qPCR primer specificity and the presence of un- or misassigned sequences in the sequencing data. Figure 6 shows the results of the Illumina sequencing at genus level. The distribution on the phylum and family level is displayed in the supporting information (Supplementary Figures 2 and 3).

When looking at the microbiota composition, it is recognizable that the composition of the fecal sample from donors B and C were quite similar and in line with a microbiota composition to be found in healthy humans dominated by Firmicutes and Bacteroidetes. 43 The composition of the fecal sample from donor A on the other hand was very different, which was confirmed by PCA at OTU level (≥97% sequence identity, Supplementary Figure 4). In particular, it had a very high abundance of Firmicutes, but low Bacteroidetes (Supplementary Figure 2). This initial fecal sample also showed the biggest change in microbiota composition investigated by Illumina sequencing after 48 h of incubation with all substrates (Supplementary Figure 4), with a big increase in Bacteroidetes and Proteobacteria at the expense of Firmicutes at the phylum level (Supplementary Figure 2). Only a few genera showed a major increase in relative abundance after incubation (Bacteroides, Clostridium XIVa, Acidaminococcus, Parabacteroides, Veillonella, Burkholderiales and Escherichia/Shigella; Figure 6). Microbiota compositions between samples that were incubated with different phenolic substrates for 48 h were very similar within the donors (Figure 6, Supplementary Figure 4). This is in agreement with the results from the qPCR and SCFA analysis. Usually, α-diversity is used to express the mean variation of species to be found in a certain microbiome. In this study, α-diversity was similar in all three fecal samples and remained high after incubation with the fecal sample from donor C, but was lower after incubation with the other two donors, particularly with donor A (Supplementary Table 4). β-Diversity could not be calculated due to the limited number of samples. PCA analysis and a dendogram were used for a visual presentation (Supplementary Figure 4).

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In a study published by Duda-Chodak ⁴³, it was concluded that flavonoid aglycones may inhibit growth of some intestinal bacteria, consequently leading to a modulation of the whole intestinal microbiome. Especially in the fermentations with the fecal sample from donor A, where the microbiota composition changes most between 0 h and 48 h, the aglycone was always present for longer than in the fermentations with

the fecal sample from donors B and C. Consequently, it might be possible that this is a reason for the bigger differences. However, qPCR and SCFA data (**Figure 5**) indicate that the control samples without phenolic substances had a similar microbiota composition and activity. The incubations of fecal microbiota from donor A showed a microbiota shift that may indicate exposure to oxygen as oxygen tolerant bacteria were stimulated (in particular *Escherichia/Shigella*). While this is principally possible, there was no indication that this actually happened in the corresponding incubations. Anaerobic conditions were checked with the addition of resazurin (0.1%) to the fermentation media. Resazurin is a redox indicator that changes from colorless via pink into blue when oxygen is present. In all three fermentations no color change was detected, suggesting that the microbiota shift seen was due to unusual initial composition of fecal sample A rather than an experimental mistake.

For decades, it is controversially discussed that manipulation of the composition by prebiotics might help to improve health. A high amount of bifidobacteria in the gut is often associated with health promoting effects. In a review published by Duenas, Munoz-Gonzalez, Cueva, Jimenez-Giron, Sanchez-Patan, Santos-Buelga, Moreno-Arribas and Bartolome studies are described, where a stimulation of the growth of beneficial bacteria, such as bifidobacteria, was caused by polyphenols, thus, exerting prebiotic-like effects. In the present study, an increase in the amount of bifidobacteria between the initial fecal samples and the fermentation samples after 48 h were detected for nearly all substrates tested. However, a significant difference of the amount of bifidobacteria between the 'control samples I' and the fermentation samples based on the qPCR results were not detected (Supplementary Table 3). So, changes are probably caused by the fermentation media or reaction conditions.

During the fermentation of carbohydrates, where SCFA often result as main products, specific gases are produced, as well. In this context, it could be also possible to cluster the bacteria based on their gas release depending on the specific glycosylated phenolic compounds. For that purpose, gas sensing is an alternative technology for measuring gases from *in vitro* fecal sample fermentation.⁴⁵ However, this was not possible to test within this study.

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Interindividual comparison of the phenolic conversion rates of the different donors. For all substrates tested, phenolic conversion rate of donor A was slower than the fermentation rates with fecal samples of donors B and C, which were in turn quite similar. For evaluating kinetics of the degradation, D₅₀ (degradation₅₀) values of all substrates were used. These represent the time point, when 50% of the substrate was totally degraded by the gut microbiota (Tables 1 and 2). It was not possible to estimate such values for the metabolites, because their formation and a possible follow-up degradation of a metabolite can take place in parallel. In general, deglycosylation reactions occurred more quickly than the breakdown of the aglycone structures. Consequently, based on the conversion rates, the metabolite profiles and the corresponding time courses differed between the different donors and fermentations, with the fecal samples from donor A showing in general less metabolites. These differences are probably caused by the individual microbial compositions in the microbiota from each donor. In the study published by Justesen, Arrigoni, Larsen and Amado 34, a major decrease of rutin was observed at first after 8 h of incubation. They concluded that it might be possible that other compounds being in the media or feces are more preferred substrates (such as carbohydrates) and are easier to ferment for the bacteria and thus, that the microorganisms probably needed an adaptation period. Moreover, Cassidy and Minihane describe wide

interindividual variability in the bioconversion of flavonoids being attributed to specific enterotypes, suggesting that individuals may be either weak or strong flavonoid converters.

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When comparing the composition of the microbiota after 48 h of incubation with the substrate T in the present study, it is noticeably that Firmicutes were more dominant in the fecal sample from donor B and C. In conjunction with above, the microbial profile of the fermentation A showed more members of the Escherichia/Shigella cluster (Figure 6). So far, more strains from the Firmicutes have been identified to be responsible for flavonoid conversion than species from Proteobacteria where Escherichia/Shigella belongs to. 18 For example, three Ruminococcaceae species and different Lachnospiraceae species were described in the mentioned review as being part of the degradation of some flavonoid structures, whereby the latter were mainly found to carry out deglycosylation reactions. Looking at the family level of the Illumina sequencing results of the present study (Supplementary Figure 3), the amount of Ruminococcaceae and Lachnospiraceae were much higher in donors B and C than in donor A. Furthermore, some species within the Lachnospiraceae family were also identified being able to catalyze reduction reactions, e.g., of p-CA.37, 38 The sequencing data do not allow resolution at species level, but it is possible that donors B and C carry more bacteria that are able to perform deglycosylations and reduction reactions and therefore result in a faster conversion rate than donor A.

Comparison of the degradation of *C*- and *O*-glycosides by human fecal microbiota. When comparing the time courses of the deglycosylations of *O*- and *C*-glycosides, the present study showed that *O*-glycosides are metabolized faster than *C*-glycosides, while the degradation rates of the two aglycone structures are very similar within a fermentation (**Tables 1 and 2**). Looking at the fermentations of T and

V with the fecal sample from donor B, exemplarily, the concentration of the Cglycoside V remained nearly constant up to 7 h and V was then fully metabolized (**Table 2**), whereas the O-glycoside T was already totally degraded within the first 7h (Table 1, Figure 3B). Additionally, with regard to the different vitexin derivatives, it was shown that the release of the O-glycoside was faster than the degradation of the intermediate product V, where the sugar is bound C-glycosidically (**Table 2**). This may be due to the fact that O-glycosides occur more frequently in the nature than Cglycosides and that microorganisms are more adapted to cleave the C-O-bond providing more binding energy. Furthermore, it can be assumed that after the consumption of C- and O-glycosides they reach different areas of the gut due to a described enhanced stability of C-glycosides over the O-glycosides. Based on the different prevailing conditions which are found for several intestinal areas, it might be also possible that C- and O-glycosides in humans are converted by different microorganisms. Braune and Blaut ²³ investigated the potential of the deglycosylations of different flavonoid C- and O-glycosides by Eubacterium cellulosolvens which was only able to cleave the C-glycoside bond in two out of seven compounds tested. In their study, both aglycones were then not further degraded to phenolic acids. In contrast, the incubation of E. cellulosolvens with six different O-glycosides showed that the microorganism was able to deglycosylate five substrates.²³ In all in vitro fermentations with the different flavonoid C- and O-glycosides, more

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than one degradation step was identified. As a consequence, not only deglycosylations occurred, but also a transformation of the aglycones to smaller phenolic compounds took place. It is described in the literature that the hydrolysis of (flavonoid) glycosides to their aglycones results in potentially more bioactive

metabolites compared to the initial compounds. However, a further microbial degradation of the aglycone to smaller metabolites can lead to the formation of more or less active compounds. 47 The fact that microorganisms are often not able to carry out all degradation steps was already described for the small phenolic acid caffeic acid by Peppercorn and Goldman 48, in 1971. This seems to be valid for the degradation of flavonoids as well. For example, in the study published by Braune and Blaut ²³, E. cellulosolvens was only able to carry out deglycosylation reactions, resulting in the aglycone structure as the only metabolite. However, in the substrates metabolized by *E. cellulosolvens*, the glucose moiety was bound at the C6-position. When glucose was present at the C8-position (e.g., vitexin) degradation did not occur. Braune and Blaut further showed that the intestinal Lachnospiraceae strain CG19-1, in contrast to E. cellulosolvens, was able to convert six out of the seven tested C-glycosides via the aglycone structure to small phenolic acids, indicating deglycosylation, ring fission, and dehydroxylation reactions.²³ Investigations done by Nakamura, et al. ⁴⁹ on the cleavage of the *C*-glycosyl bond of puerarin (an isoflavone glucoside) by a strain called PUE showed only a conversion to its aglycone daidzein. Smaller phenolic acids were not detected. 49 In this context, it is not clear whether the degradation steps of the substrates used in the present study are carried out by only one or even more microorganisms. In summary, all substrates used in the present in vitro study were converted yielding lower molecular weight phenolic acids. Deglycosylation reactions were carried out by the microbiota first, followed by the further breakdown of the aglycone structure.

Comparing the donors with each other, similar metabolites were detected, even

though different time courses for metabolite formation were observed. With regard to

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the microbial composition or activity (SCFA), an influence of the flavonoid glycosides was not detected.

When looking at the results of the present study, the degradation of the selected phenolic compounds showed more or less the same course of degradation that resulted in very similar metabolite profiles, as well. Only the fermentation rates were different. Therefore, we did not expect very different degradation pathways and more outcomes when fermenting the substrates with fecal samples of additional volunteers. The degradation pathways might only be significantly different, when using fecal samples of volunteers having a known disorder (e.g., by taking antibiotics) or significantly different dietary habits. It is known that one donor of the present study was a vegetarian, but significant differences in metabolite formation were not detected, at all. Also, seasonable differences might be possible, too, but the main aim of the study was to monitor the degradation profile with regard to the (complex) phenolic compounds selected. Therefore, more fermentations with more fecal samples were not carried out. When comparing with *in vitro* fermentation experiments described in the literature it is obvious that the use of one to four different fecal samples is usual.^{21, 29-31}

In conclusion, flavonoid glycosides can be metabolized by the human gut microorganisms when consumed in concentrations found in typical diets (~20 mg/day), but a prebiotic effect of the phenolic compounds seems not to be achievable. However, in a review by Etxeberria, et al. ⁵⁰, *in vitro* and *in vivo* studies are described which showed an influence of phenolic compounds on the microbiota composition. The effects were often only significant when the concentrations were artificially high. This does not mean that a certain influence of the microbiome is not possible, as not all species present in the gut can be analyzed comprehensively to

date. But, it is also possible that a substrate on its own is not able to influence the gut microbiota and the described effects on the microbiota composition are a result of the interaction of different phenolic compounds and other food components being present in the food matrix such as proteins. Besides other research groups, Ozdal, et al. ⁵¹ described that a polyphenol-protein interaction results in changes in the structural, functional, and nutritional properties of both compounds. This interaction is influenced by several parameters such as pH, temperature, and the chemical structure of the compounds. Therefore, interactive effects of the polyphenols with further compounds might influence their metabolism by the microbiota to the individual strength/binding mechanism of the polyphenol-protein interaction, when consuming whole foods.

Microbial transformation products might affect microbial composition and might therefore lead to different consequences for the host and its health.^{21, 52} Such an effect could be more pronounced after consumption of *C*-glycosides, as their slower cleavage leads to a prolonged exposure to the intact flavonoid. Moreover, an impact on deeper regions of the colon seems to be possible due to the enhanced stability. With regard to diversity, it can be hypothesized that the more complex the substrates (for gastrointestinal fermentation) are, more different metabolites can be formed likely resulting in a higher variation of gut microbial composition. Although this research topic is now studied for more than two decades, a final proof of a health-beneficial effect of isolated phenolic compounds is still missing. In the future, research has to be extended towards synergisms/interactions with other food compounds and all what we would call food matrix.

Abbreviations. AP, apigenin; p-CA, *p*-coumaric acid; 4-HPPA, 3-(4-K, kaempferol; K-DG-R, kaempferol-3.4'-Ohydroxyphenyl)propionic acid; diglucoside-7-O-rhamnoside; K-G, kaempferol-3-O-glucoside; LOD, limit of detection; PAA, phenylacetic acid; PCA, Principal Component Analysis; 3-PPA, 3phenylpropionic acid; SCFA, short chain fatty acids; T, tiliroside; V, vitexin; V-G, vitexin-2"-O-glucoside; V-R, vitexin-2"-O-rhamnoside; V-X, vitexin-2"-O-xyloside.

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SUPPORTING INFORMATION

Isolation method of vitexin-2"-O-glucoside and vitexin-2"-O-xyloside (material and methods), Illumina sequencing of 16S rRNA genes (material and methods), figures about the relative abundance of OTUs grouped at phylum and family level (results), PCA with the sequencing data (results), tables about the SCFA production, microbiota profile and Illumina sequencing operational taxonomic units (results). This material is available free of charge via the Internet at http://pubs.acs.org

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LIST OF FIGURE CAPTIONS

- Figure 1: Structures of the compounds used for the *in vitro* fermentations.
- Figure 2: Overview of the degradation pattern of the kaempferol-O-derivates und
- apigenin-C-derivatives used in the in vitro study (AP, apigenin; p-CA, p-coumaric
- 849 acid; Glc, glucose; 4-HPAA, 3-(4-hydroxyphenyl)acetic acid; 4-HPPA, 3-(4-
- 850 hydroxyphenyl)propionic acid; K, kaempferol; K-DG-R, kaempferol-3,4'-O-
- 851 diglucoside-7-O-rhamnoside; K-G, kaempferol-3-O-glucoside; PAA, phenylacetic
- acid; 3-PPA, 3-phenylpropionic acid; Rha, rhamnose; T, tiliroside; V, vitexin; V-G,
- vitexin-2"-O-glucoside; V-R, vitexin-2"-O-rhamnoside; V-X, vitexin-2"-O-xyloside; - -
- > = pathway is not totally identified).
- 855 Figure 3: In vitro degradation and metabolite formation of tiliroside with three
- 856 different donors (A, B, and C). Data represent the mean and standard deviation of
- triplicates (T, tiliroside; K-G, kaempferol-3-O-glucoside; K, kaempferol; 4-HPPA, 3-(4-
- 858 hydroxyphenyl)propionic acid; p-CA, p-coumaric acid; 3-PPA, 3-phenylpropionic
- 859 acid).
- Figure 4: Comparison of vitexin formation within the in vitro fermentations with
- different vitexin derivatives of donor B. Data are normalized to 100%.
- Figure 5: Principal component analysis with the percentage data for net SCFA and
- production microbiota composition (qPCR) after 48 h of incubation. Colors are coded
- by different donor (donor A: black, donor B: red, donor C: blue).
- Figure 6: Relative abundance of OTUs obtained by Illumina sequencing of the three
- different initial fecal samples (donor A, B and C) and after 48 h of incubation with
- different substrates. OTUs are grouped together at the genus level.

TABLES

Table 1: Concentration of the substrates and metabolites within the *in vitro* degradation of K, T, and K-DG-R with the three different donors (A, B, and C). Data represent the mean and standard deviation of triplicates (D₅₀, time point when 50% of the substrate was degraded).

		Concentration [µM]					
Compound	Donor	0 h	3 h	7 h	24 h	48 h	
Kaempferol	А	232 ± 16.0	201 ± 15.8	179 ± 51.4	170 ± 8.32	34.0 ± 20.7	33 h 32 min
	В	68.9 ± 13.4	61.1 ± 32.9	<lod< td=""><td><lod< td=""><td><lod< td=""><td>4 h 45 min</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>4 h 45 min</td></lod<></td></lod<>	<lod< td=""><td>4 h 45 min</td></lod<>	4 h 45 min
	С	27.5 ± 5.31	16.9 ± 4.53	8.28 ± 11.8	<lod< td=""><td><lod< td=""><td>4 h 28 min</td></lod<></td></lod<>	<lod< td=""><td>4 h 28 min</td></lod<>	4 h 28 min
K metabolites							
4-HPPA	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>32.8 ± 0.0100</td><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>32.8 ± 0.0100</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>32.8 ± 0.0100</td><td></td></lod<></td></lod<>	<lod< td=""><td>32.8 ± 0.0100</td><td></td></lod<>	32.8 ± 0.0100	
	В	<lod< td=""><td><lod< td=""><td>19.4*</td><td>5.97*</td><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td>19.4*</td><td>5.97*</td><td><lod< td=""><td></td></lod<></td></lod<>	19.4*	5.97*	<lod< td=""><td></td></lod<>	
	С	<lod< td=""><td><lod< td=""><td>13.8 ± 5.91**</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td>13.8 ± 5.91**</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	13.8 ± 5.91**	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
PAA	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td>13.7 ± 4.91</td><td>21.1 ± 18.6</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>13.7 ± 4.91</td><td>21.1 ± 18.6</td><td></td></lod<></td></lod<>	<lod< td=""><td>13.7 ± 4.91</td><td>21.1 ± 18.6</td><td></td></lod<>	13.7 ± 4.91	21.1 ± 18.6	
Tiliroside	Α	105 ± 13.2	84.4 ± 6.65	54.6 ± 13.0	<lod< td=""><td><lod< td=""><td>7 h 17 min</td></lod<></td></lod<>	<lod< td=""><td>7 h 17 min</td></lod<>	7 h 17 min
	В	183 ± 48.5	125 ± 35.1	<lod< td=""><td><lod< td=""><td><lod< td=""><td>4 h 4 min</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>4 h 4 min</td></lod<></td></lod<>	<lod< td=""><td>4 h 4 min</td></lod<>	4 h 4 min
	С	107 ± 52.7	75.7 ± 38.2	<lod< td=""><td><lod< td=""><td><lod< td=""><td>3 h</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>3 h</td></lod<></td></lod<>	<lod< td=""><td>3 h</td></lod<>	3 h
T metabolites							
K-G	Α	5.85 ± 2.41	7.16 ± 0.522	8.91 ± 2.62	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	В	91.4 ± 28.1	22.1 ± 10.5	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	31.7 ± 6.36	16.8 ± 3.82	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
K	Α	32.3 ± 17.7	95.3 ± 21.2	145 ± 19.9	89.6 ± 0.719	79.4 ± 27.2	
	В	5.24 ± 0.176	39.8 ± 8.89	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	4.53 ± 2.33	8.70 ± 4.45	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
p-CA	Α	19.5 ± 2.61	31.3 ± 12.2	123 ± 19.5	17.1 ± 24.4	3.02 ± 0.776	
	В	3.97 ± 0.453	5.86 ± 1.46	12.7 ± 8.36	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	8.14 ± 1.56	19.4 ± 4.35	49.1 ± 9.71	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	

4-HPPA	A B	<lod <lod< th=""><th><lod <lod< th=""><th><lod 99.9 ± 53.2</lod </th><th>123 ± 3.56 <lod< th=""><th>121 ± 12.4 <lod< th=""></lod<></th></lod<></th></lod<></lod </th></lod<></lod 	<lod <lod< th=""><th><lod 99.9 ± 53.2</lod </th><th>123 ± 3.56 <lod< th=""><th>121 ± 12.4 <lod< th=""></lod<></th></lod<></th></lod<></lod 	<lod 99.9 ± 53.2</lod 	123 ± 3.56 <lod< th=""><th>121 ± 12.4 <lod< th=""></lod<></th></lod<>	121 ± 12.4 <lod< th=""></lod<>
	C	<lod <lod< td=""><td><lod <lod< td=""><td>46.9 ± 19.1</td><td><lod <lod< td=""><td><lod <lod< td=""></lod<></lod </td></lod<></lod </td></lod<></lod </td></lod<></lod 	<lod <lod< td=""><td>46.9 ± 19.1</td><td><lod <lod< td=""><td><lod <lod< td=""></lod<></lod </td></lod<></lod </td></lod<></lod 	46.9 ± 19.1	<lod <lod< td=""><td><lod <lod< td=""></lod<></lod </td></lod<></lod 	<lod <lod< td=""></lod<></lod
3-PPA	A	<lod< td=""><td><lod <lod< td=""><td>40.9 <u>1</u> 19.1 <lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<></lod </td></lod<>	<lod <lod< td=""><td>40.9 <u>1</u> 19.1 <lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<></lod 	40.9 <u>1</u> 19.1 <lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>27.6 ± 26.8**</td><td>5.36 ± 9.28**</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>27.6 ± 26.8**</td><td>5.36 ± 9.28**</td></lod<></td></lod<>	<lod< td=""><td>27.6 ± 26.8**</td><td>5.36 ± 9.28**</td></lod<>	27.6 ± 26.8**	5.36 ± 9.28**
	C	<lod< td=""><td><lod< td=""><td><lod< td=""><td>91.8 ± 13.3**</td><td>38.0 ± 65.8**</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>91.8 ± 13.3**</td><td>38.0 ± 65.8**</td></lod<></td></lod<>	<lod< td=""><td>91.8 ± 13.3**</td><td>38.0 ± 65.8**</td></lod<>	91.8 ± 13.3**	38.0 ± 65.8**
K-DG-R	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
K-DG-R metabolites						
K-G	Α	8.77 ± 1.94	36.6 ± 13.2	10.3 ± 1.17	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
	В	35.8 ± 6.55	88.7 ± 61.2	<lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
	С	3.92 ± 0.976	30.7 ± 24.9	<lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
K	Α	13.0 ± 7.57	131 ± 12.1	170 ± 14.4	151 ± 40.8	6.60 ± 1.65
	В	<lod< td=""><td>54.9 ± 0.701</td><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<>	54.9 ± 0.701	<lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
	С	<lod< td=""><td>13.3 ± 5.25</td><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<>	13.3 ± 5.25	<lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
4-HPPA	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>33.5 ± 2.88</td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>33.5 ± 2.88</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>33.5 ± 2.88</td></lod<></td></lod<>	<lod< td=""><td>33.5 ± 2.88</td></lod<>	33.5 ± 2.88
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
PAA	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>21.8 ± 23.4</td><td>15.5 ± 19.9</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>21.8 ± 23.4</td><td>15.5 ± 19.9</td></lod<></td></lod<>	<lod< td=""><td>21.8 ± 23.4</td><td>15.5 ± 19.9</td></lod<>	21.8 ± 23.4	15.5 ± 19.9
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td>3.05 ± 7.74</td><td>18.3 ± 10.1</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>3.05 ± 7.74</td><td>18.3 ± 10.1</td></lod<></td></lod<>	<lod< td=""><td>3.05 ± 7.74</td><td>18.3 ± 10.1</td></lod<>	3.05 ± 7.74	18.3 ± 10.1

<LOD (limit of detection)</p>
* the metabolite was only detected in one sample out of the triplicate fermentation
** the metabolite was only detected in two samples out of the triplicate fermentation

Table 2: Concentrations of the substrates and metabolites within the *in vitro* degradation of AP, V, V-G, V-R and V-X with the three different donors (A, B, and C). Data represent the mean and standard deviation of triplicates (D₅₀, time point when 50% of the substrate was degraded).

		Concentration [µM]					
Compound	Donor	0 h	3 h	7 h	24 h	48 h	
Apigenin	А	106 ± 16.0	97.9 ± 7.57	98.6 ± 11.7	65.4 ± 26.9	34.0 ± 20.7	33 h 29 min
	В	22.9 ± 14.5	15.4 ± 3.02	<lod< td=""><td><lod< td=""><td><lod< td=""><td>4 h 35 min</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>4 h 35 min</td></lod<></td></lod<>	<lod< td=""><td>4 h 35 min</td></lod<>	4 h 35 min
	С	24.1 ± 2.66	25.3 ± 6.55	<lod< td=""><td><lod< td=""><td><lod< td=""><td>5 h 6 min</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>5 h 6 min</td></lod<></td></lod<>	<lod< td=""><td>5 h 6 min</td></lod<>	5 h 6 min
AP metabolites							
4-HPPA	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td>38.7 ± 9.56</td><td>173 ± 55.5</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>38.7 ± 9.56</td><td>173 ± 55.5</td><td></td></lod<></td></lod<>	<lod< td=""><td>38.7 ± 9.56</td><td>173 ± 55.5</td><td></td></lod<>	38.7 ± 9.56	173 ± 55.5	
	В	<lod< td=""><td><lod< td=""><td>187 ± 66.2</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td>187 ± 66.2</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	187 ± 66.2	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	<lod< td=""><td>32.9 ± 18.6</td><td>124 ± 7.95</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	32.9 ± 18.6	124 ± 7.95	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
3-PPA	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>8.01 ± 1.55</td><td>49.8 ± 35.1</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>8.01 ± 1.55</td><td>49.8 ± 35.1</td><td></td></lod<></td></lod<>	<lod< td=""><td>8.01 ± 1.55</td><td>49.8 ± 35.1</td><td></td></lod<>	8.01 ± 1.55	49.8 ± 35.1	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td>143 ± 24.1</td><td>214 ± 39.4**</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>143 ± 24.1</td><td>214 ± 39.4**</td><td></td></lod<></td></lod<>	<lod< td=""><td>143 ± 24.1</td><td>214 ± 39.4**</td><td></td></lod<>	143 ± 24.1	214 ± 39.4**	
PAA	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>13.8 ± 12.1</td><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>13.8 ± 12.1</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>13.8 ± 12.1</td><td></td></lod<></td></lod<>	<lod< td=""><td>13.8 ± 12.1</td><td></td></lod<>	13.8 ± 12.1	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>24.4 ± 21.3</td><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>24.4 ± 21.3</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>24.4 ± 21.3</td><td></td></lod<></td></lod<>	<lod< td=""><td>24.4 ± 21.3</td><td></td></lod<>	24.4 ± 21.3	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td>52.4 ± 14.9</td><td>50.4 ± 7.64</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>52.4 ± 14.9</td><td>50.4 ± 7.64</td><td></td></lod<></td></lod<>	<lod< td=""><td>52.4 ± 14.9</td><td>50.4 ± 7.64</td><td></td></lod<>	52.4 ± 14.9	50.4 ± 7.64	
Vitexin	Α	113 ± 23.1	117 ± 27.3	108 ± 11.6	112 ± 1.04	<lod< td=""><td>35 h 52 min</td></lod<>	35 h 52 min
	В	147 ± 21.9	96.6 ± 31.1	115 ± 17.8	<lod< td=""><td><lod< td=""><td>13 h 8 min</td></lod<></td></lod<>	<lod< td=""><td>13 h 8 min</td></lod<>	13 h 8 min
	B C	131 ± 4.53	121*	128 ± 4.85	<lod< td=""><td><lod< td=""><td>15 h 17 min</td></lod<></td></lod<>	<lod< td=""><td>15 h 17 min</td></lod<>	15 h 17 min
V metabolites							
AP	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>39.4 ± 30.5</td><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>39.4 ± 30.5</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>39.4 ± 30.5</td><td></td></lod<></td></lod<>	<lod< td=""><td>39.4 ± 30.5</td><td></td></lod<>	39.4 ± 30.5	
		<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	B C	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
4-HPPA	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td>22.6 ± 0.961</td><td>102 ± 32.6</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>22.6 ± 0.961</td><td>102 ± 32.6</td><td></td></lod<></td></lod<>	<lod< td=""><td>22.6 ± 0.961</td><td>102 ± 32.6</td><td></td></lod<>	22.6 ± 0.961	102 ± 32.6	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>123 ± 6.76</td><td>80.2 ± 17.9</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>123 ± 6.76</td><td>80.2 ± 17.9</td><td></td></lod<></td></lod<>	<lod< td=""><td>123 ± 6.76</td><td>80.2 ± 17.9</td><td></td></lod<>	123 ± 6.76	80.2 ± 17.9	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
3-PPA	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>41.1 ± 27.0**</td><td>25.7 ± 4.90*</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>41.1 ± 27.0**</td><td>25.7 ± 4.90*</td><td></td></lod<></td></lod<>	<lod< td=""><td>41.1 ± 27.0**</td><td>25.7 ± 4.90*</td><td></td></lod<>	41.1 ± 27.0**	25.7 ± 4.90*	

	C	<lod< td=""><td><lod< td=""><td><lod< td=""><td>131 ± 30.7</td><td>142 ± 35.9</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>131 ± 30.7</td><td>142 ± 35.9</td><td></td></lod<></td></lod<>	<lod< td=""><td>131 ± 30.7</td><td>142 ± 35.9</td><td></td></lod<>	131 ± 30.7	142 ± 35.9	
PAA	Α	<lod< td=""><td><lod< td=""><td>6.70*</td><td><lod< td=""><td>10.9 ± 4.79</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td>6.70*</td><td><lod< td=""><td>10.9 ± 4.79</td><td></td></lod<></td></lod<>	6.70*	<lod< td=""><td>10.9 ± 4.79</td><td></td></lod<>	10.9 ± 4.79	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>43.8 ± 10.2</td><td>47.0 ± 10.2</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>43.8 ± 10.2</td><td>47.0 ± 10.2</td><td></td></lod<></td></lod<>	<lod< td=""><td>43.8 ± 10.2</td><td>47.0 ± 10.2</td><td></td></lod<>	43.8 ± 10.2	47.0 ± 10.2	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td>47.9 ± 19.8</td><td>47.3 ± 9.75</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>47.9 ± 19.8</td><td>47.3 ± 9.75</td><td></td></lod<></td></lod<>	<lod< td=""><td>47.9 ± 19.8</td><td>47.3 ± 9.75</td><td></td></lod<>	47.9 ± 19.8	47.3 ± 9.75	
Vitexin-glucoside	Α	1.84 ± 0.342	2.02 ± 0.652	2.34 ± 0.519	1.22 ± 0.765	0.769 ± 0.152	27 h 50 min
	В	0.500 ± 0.200	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>1 h 29 min</td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>1 h 29 min</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>1 h 29 min</td></lod<></td></lod<>	<lod< td=""><td>1 h 29 min</td></lod<>	1 h 29 min
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>-</td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>-</td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>-</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>-</td></lod<></td></lod<>	<lod< td=""><td>-</td></lod<>	-
V-G metabolites							
V	Α	<lod< td=""><td>0.351 ± 0.0137</td><td>0.714 ± 0.127</td><td>1.29 ± 0.612</td><td><lod< td=""><td></td></lod<></td></lod<>	0.351 ± 0.0137	0.714 ± 0.127	1.29 ± 0.612	<lod< td=""><td></td></lod<>	
	В	<lod< td=""><td>7.02 ± 1.01</td><td>3.52 ± 2.09</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	7.02 ± 1.01	3.52 ± 2.09	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	C	0.824 ± 0.0560	2.29 ± 0.578	2.45 ± 1.23	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
4-HPPA	Ä	<lod< td=""><td><lod< td=""><td><lod< td=""><td>25.9*</td><td>37.5 ± 14.4**</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>25.9*</td><td>37.5 ± 14.4**</td><td></td></lod<></td></lod<>	<lod< td=""><td>25.9*</td><td>37.5 ± 14.4**</td><td></td></lod<>	25.9*	37.5 ± 14.4**	
711117	В	<lod <lod< td=""><td><lod< td=""><td>14.0 ± 3.79**</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></lod 	<lod< td=""><td>14.0 ± 3.79**</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	14.0 ± 3.79**	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	<lod <lod< td=""><td><lod <lod< td=""><td>14.0 ± 3.79 <lod< td=""><td><lod <lod< td=""><td><lod <lod< td=""><td></td></lod<></lod </td></lod<></lod </td></lod<></td></lod<></lod </td></lod<></lod 	<lod <lod< td=""><td>14.0 ± 3.79 <lod< td=""><td><lod <lod< td=""><td><lod <lod< td=""><td></td></lod<></lod </td></lod<></lod </td></lod<></td></lod<></lod 	14.0 ± 3.79 <lod< td=""><td><lod <lod< td=""><td><lod <lod< td=""><td></td></lod<></lod </td></lod<></lod </td></lod<>	<lod <lod< td=""><td><lod <lod< td=""><td></td></lod<></lod </td></lod<></lod 	<lod <lod< td=""><td></td></lod<></lod 	
PAA	A	<lod <lod< td=""><td><lod <lod< td=""><td><lod <lod< td=""><td><lod <lod< td=""><td>7.37 ± 1.30**</td><td></td></lod<></lod </td></lod<></lod </td></lod<></lod </td></lod<></lod 	<lod <lod< td=""><td><lod <lod< td=""><td><lod <lod< td=""><td>7.37 ± 1.30**</td><td></td></lod<></lod </td></lod<></lod </td></lod<></lod 	<lod <lod< td=""><td><lod <lod< td=""><td>7.37 ± 1.30**</td><td></td></lod<></lod </td></lod<></lod 	<lod <lod< td=""><td>7.37 ± 1.30**</td><td></td></lod<></lod 	7.37 ± 1.30**	
PAA							
	В	<lod< td=""><td><lod< td=""><td><lod.< td=""><td>38.3 ± 10.0</td><td>32.2 ± 3.6</td><td></td></lod.<></td></lod<></td></lod<>	<lod< td=""><td><lod.< td=""><td>38.3 ± 10.0</td><td>32.2 ± 3.6</td><td></td></lod.<></td></lod<>	<lod.< td=""><td>38.3 ± 10.0</td><td>32.2 ± 3.6</td><td></td></lod.<>	38.3 ± 10.0	32.2 ± 3.6	
	С	<lod< td=""><td><lod< td=""><td>.n.d</td><td>35.1 ± 29.0</td><td>19.6 ± 1.41</td><td></td></lod<></td></lod<>	<lod< td=""><td>.n.d</td><td>35.1 ± 29.0</td><td>19.6 ± 1.41</td><td></td></lod<>	.n.d	35.1 ± 29.0	19.6 ± 1.41	
Vitexin-rhamnoside	Α	8.43 ± 0.623	8.95 ± 1.38	12.3 ± 3.46	13.7 ± 1.66	<lod< td=""><td>36 h 3 min</td></lod<>	36 h 3 min
	В	12.0 ± 5.17	9.71 ± 0.424	4.65 ± 0.443	<lod< td=""><td><lod< td=""><td>5 h 55 min</td></lod<></td></lod<>	<lod< td=""><td>5 h 55 min</td></lod<>	5 h 55 min
	С	5.58 ± 1.43	7.51 ± 1.40	10.3 ± 0.550	<lod< td=""><td><lod< td=""><td>15 h 29 min</td></lod<></td></lod<>	<lod< td=""><td>15 h 29 min</td></lod<>	15 h 29 min
V-R metabolites							
V	Α	<lod< td=""><td><lod< td=""><td>2.29 ± 0.851</td><td>10.7 ± 7.45</td><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td>2.29 ± 0.851</td><td>10.7 ± 7.45</td><td><lod< td=""><td></td></lod<></td></lod<>	2.29 ± 0.851	10.7 ± 7.45	<lod< td=""><td></td></lod<>	
	В	<lod< td=""><td>1.65 ± 0.415</td><td>68.8 ± 19.8</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	1.65 ± 0.415	68.8 ± 19.8	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	0.845 ± 0.154	1.41 ± 0.475	7.82 ± 1.30	0.748 ± 0.063	<lod< td=""><td></td></lod<>	
AP	Ā	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>28.2*</td><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>28.2*</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>28.2*</td><td></td></lod<></td></lod<>	<lod< td=""><td>28.2*</td><td></td></lod<>	28.2*	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	C	<lod< td=""><td><lod< td=""><td><lod< td=""><td>18.8*</td><td>16.6*</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>18.8*</td><td>16.6*</td><td></td></lod<></td></lod<>	<lod< td=""><td>18.8*</td><td>16.6*</td><td></td></lod<>	18.8*	16.6*	
4-HPPA	Ä	<lod< td=""><td><lod< td=""><td><lod< td=""><td>23.2 ± 0**</td><td>130.8 ± 48.1</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>23.2 ± 0**</td><td>130.8 ± 48.1</td><td></td></lod<></td></lod<>	<lod< td=""><td>23.2 ± 0**</td><td>130.8 ± 48.1</td><td></td></lod<>	23.2 ± 0**	130.8 ± 48.1	
	В	<lod <lod< td=""><td><lod< td=""><td><lod <lod< td=""><td>230 ± 25.7</td><td>169 ± 59.9</td><td></td></lod<></lod </td></lod<></td></lod<></lod 	<lod< td=""><td><lod <lod< td=""><td>230 ± 25.7</td><td>169 ± 59.9</td><td></td></lod<></lod </td></lod<>	<lod <lod< td=""><td>230 ± 25.7</td><td>169 ± 59.9</td><td></td></lod<></lod 	230 ± 25.7	169 ± 59.9	
	C	<lod <lod< td=""><td><lod <lod< td=""><td><lod <lod< td=""><td>230 ± 23.7 29.2*</td><td>18.0*</td><td></td></lod<></lod </td></lod<></lod </td></lod<></lod 	<lod <lod< td=""><td><lod <lod< td=""><td>230 ± 23.7 29.2*</td><td>18.0*</td><td></td></lod<></lod </td></lod<></lod 	<lod <lod< td=""><td>230 ± 23.7 29.2*</td><td>18.0*</td><td></td></lod<></lod 	230 ± 23.7 29.2*	18.0*	
2.004	A		<lod <lod< td=""><td></td><td></td><td><lod< td=""><td></td></lod<></td></lod<></lod 			<lod< td=""><td></td></lod<>	
3-PPA		<lod< td=""><td></td><td><lod< td=""><td><lod< td=""><td></td><td></td></lod<></td></lod<></td></lod<>		<lod< td=""><td><lod< td=""><td></td><td></td></lod<></td></lod<>	<lod< td=""><td></td><td></td></lod<>		
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
D.4.4	C	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>236*</td><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>236*</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>236*</td><td></td></lod<></td></lod<>	<lod< td=""><td>236*</td><td></td></lod<>	236*	
PAA	A	<lod< td=""><td><lod< td=""><td>4.84*</td><td><lod< td=""><td>127*</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td>4.84*</td><td><lod< td=""><td>127*</td><td></td></lod<></td></lod<>	4.84*	<lod< td=""><td>127*</td><td></td></lod<>	127*	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>52.4 ± 7.21</td><td>68.6 ± 22.5</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>52.4 ± 7.21</td><td>68.6 ± 22.5</td><td></td></lod<></td></lod<>	<lod< td=""><td>52.4 ± 7.21</td><td>68.6 ± 22.5</td><td></td></lod<>	52.4 ± 7.21	68.6 ± 22.5	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td>16.0 ± 11.8</td><td>41.4 ± 34.2</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>16.0 ± 11.8</td><td>41.4 ± 34.2</td><td></td></lod<></td></lod<>	<lod< td=""><td>16.0 ± 11.8</td><td>41.4 ± 34.2</td><td></td></lod<>	16.0 ± 11.8	41.4 ± 34.2	

Vitexin-xyloside***	А	3.99 ± 0.934	2.94 ± 0.194	3.88 ± 0.314	<lod< th=""><th><lod< th=""><th>15 h 16 min</th></lod<></th></lod<>	<lod< th=""><th>15 h 16 min</th></lod<>	15 h 16 min
•	В	3.23 ± 0.962	3.48 ± 0.367	4.62 ± 0.559	<lod< td=""><td><lod< td=""><td>15 h 36 min</td></lod<></td></lod<>	<lod< td=""><td>15 h 36 min</td></lod<>	15 h 36 min
	С	1.37 ± 0.336	1.19 ± 0.218	<lod< td=""><td><lod< td=""><td><lod< td=""><td>4 h 39 min</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>4 h 39 min</td></lod<></td></lod<>	<lod< td=""><td>4 h 39 min</td></lod<>	4 h 39 min
V-X metabolites							
V	Α	1.60 ± 0.284	6.20 ± 0.593	14.2 ± 2.05	57.5 ± 5.07	<lod< td=""><td></td></lod<>	
	В	<lod< td=""><td>0.746 ± 0.093</td><td>6.96 ± 2.07</td><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	0.746 ± 0.093	6.96 ± 2.07	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
	С	0.949 ± 0.034	2.81 ± 1.03	14.7 ± 5.78	0.710 ± 0.008	<lod< td=""><td></td></lod<>	
4-HPPA	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td>24.6*</td><td>60.9 ± 7.46</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>24.6*</td><td>60.9 ± 7.46</td><td></td></lod<></td></lod<>	<lod< td=""><td>24.6*</td><td>60.9 ± 7.46</td><td></td></lod<>	24.6*	60.9 ± 7.46	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>$36.8 \pm 9.47**$</td><td>32.7 ± 30.2**</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>$36.8 \pm 9.47**$</td><td>32.7 ± 30.2**</td><td></td></lod<></td></lod<>	<lod< td=""><td>$36.8 \pm 9.47**$</td><td>32.7 ± 30.2**</td><td></td></lod<>	$36.8 \pm 9.47**$	32.7 ± 30.2**	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td></td></lod<></td></lod<>	<lod< td=""><td></td></lod<>	
PAA	Α	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>9.41 ± 2.47</td><td></td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>9.41 ± 2.47</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>9.41 ± 2.47</td><td></td></lod<></td></lod<>	<lod< td=""><td>9.41 ± 2.47</td><td></td></lod<>	9.41 ± 2.47	
	В	<lod< td=""><td><lod< td=""><td><lod< td=""><td>54.0 ± 10.1</td><td>65.4 ± 20.7</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>54.0 ± 10.1</td><td>65.4 ± 20.7</td><td></td></lod<></td></lod<>	<lod< td=""><td>54.0 ± 10.1</td><td>65.4 ± 20.7</td><td></td></lod<>	54.0 ± 10.1	65.4 ± 20.7	
	С	<lod< td=""><td><lod< td=""><td><lod< td=""><td>42.2 ± 13.0</td><td>38.9 ± 13.3</td><td></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>42.2 ± 13.0</td><td>38.9 ± 13.3</td><td></td></lod<></td></lod<>	<lod< td=""><td>42.2 ± 13.0</td><td>38.9 ± 13.3</td><td></td></lod<>	42.2 ± 13.0	38.9 ± 13.3	

<LOD (limit of detection)</p>
* the metabolite was only detected in one sample out of the triplicate fermentation
** the metabolite was only detected in two samples out of the triplicate fermentation
*** Vitexin-xyloside was quantified via Vitexin-rhamnoside

FIGURE GRAPHICS

Figure 1

$$R_3$$
O
O
 R_2
OH
O
 R_2

Kaempferol: $R_1 \& R_2 \& R_3 = OH$

Kaempferol-3,4'-O-diglucoside-7-O-rhamnoside: $R_1 \& R_2$ = glucosyl, R_3 = rhamnosyl

Tiliroside: R_1 = OH, R_2 = glucosyl-6"-trans-p-coumaroyl, R_3 = OH

Apigenin

Vitexin: $R_1 = OH$

Vitexin-2"-O-glucoside: R_1 = glucosyl Vitexin-2"-O-rhamnoside: R_1 = rhamnosyl

Vitexin-2"-O-xyloside: $R_1 = xylosyl$

Figure 2

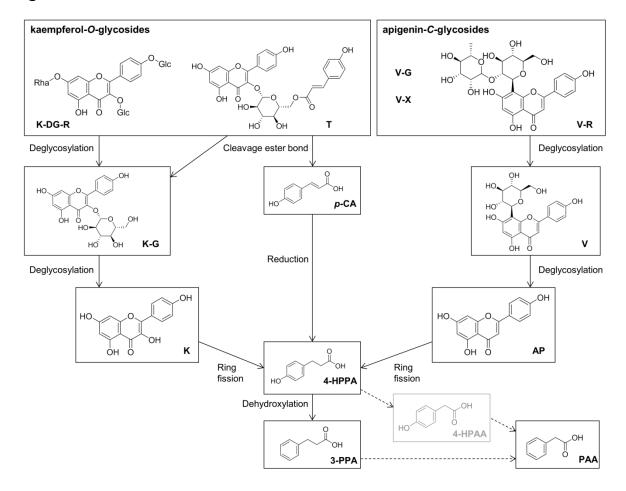
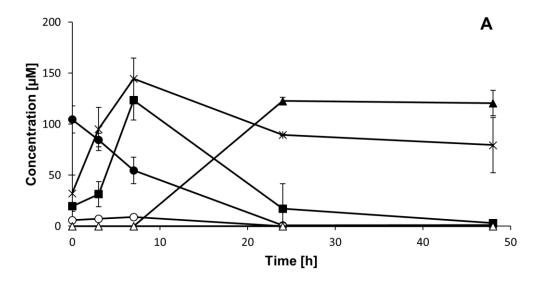
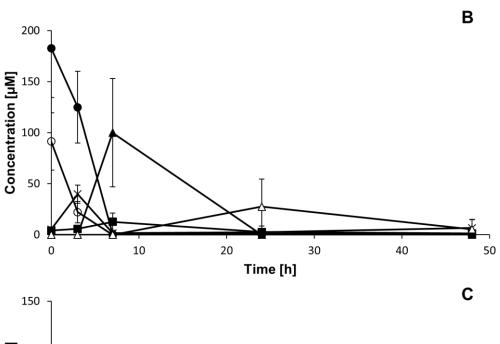


Figure 3





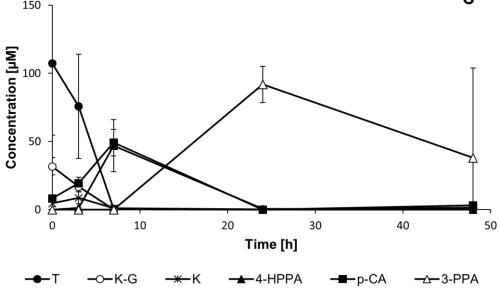


Figure 4

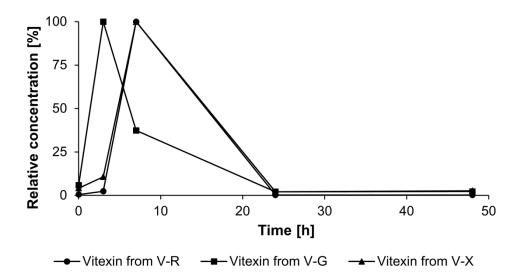


Figure 5

Bacteria% and SCFA%

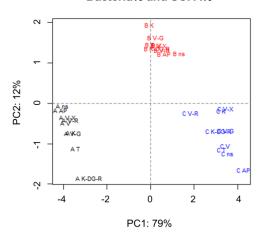


Figure 6

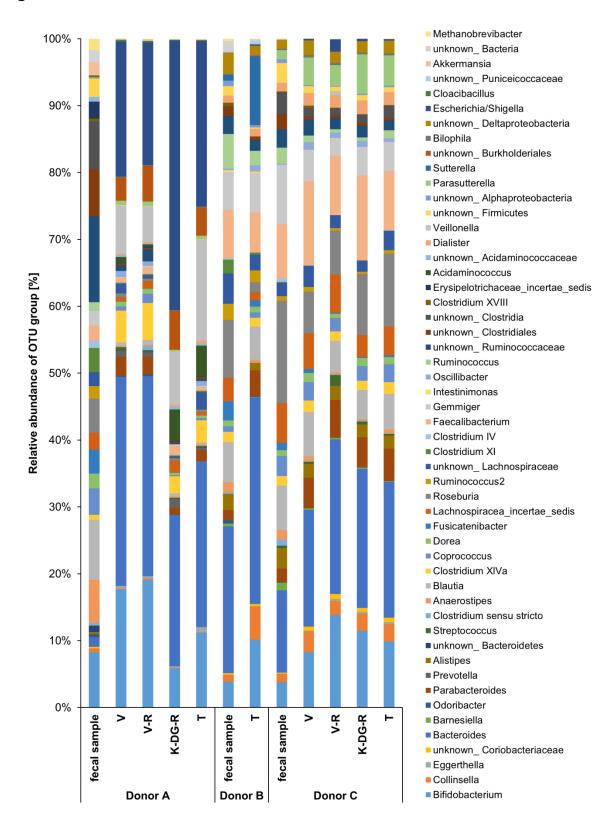


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